# Arsenic, zinc and copper accumulation in cultured milkfish from ponds using groundwater near the urban area in southwest Taiwan

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Abstract: Bioaccumulation of arsenic (As), zinc (Zn) and copper (Cu) in milkfish (*Chanos chanos*) from the culture ponds using groundwater near the urban area in southwest Taiwan was investigated. Samples of juvenile milkfish and ambient water were obtained from 8 culture ponds from the suburban area. The bioconcentration factor (BCF) of milkfish was calculated to indicate the accumulation ability of milkfish. The resulting data showed that the As, Zn and Cu concentrations in pond water were  $63.92 \pm 57.71 \ \mu g \ L^{-1}$ ,  $69.36 \pm 27.81 \ \mu g \ L^{-1}$  and  $11.09 \pm 15.22 \ \mu g \ L^{-1}$ , respectively, while in milkfish these concentrations were  $0.94 \pm 1.34 \ \mu g \ g^{-1}$ ,  $2.01 \pm 0.96 \ \mu g \ g^{-1}$  and  $40.31 \pm 17.25 \ \mu g \ g^{-1}$ , respectively. The As, Zn and Cu levels in milkfish showed significant positive relations to the As, Zn and Cu concentrations in pond water. The high BCF values of As, Zn and Cu accumulated in fish ( $12.60 \pm 4.68$ ,  $32.31 \pm 12.53$  and  $4029.04 \pm 1623.96$ , respectively) show that those aquacultural milkfish from the suburban area are contaminated by the ambient water and have a high tolerance against these pollutants. Since those fish will be sold to the markets in cities nearby, ingestion of in this way contaminated milkfish could result in overexposure of As, Zn and/or Cu in inhabitants and lead to adverse health effects. Besides the risk on human health, using groundwater for aquaculture may also have a negative effect on the land above the aquifer because it may sink and cause damage on property and fields in the suburban and/or urban areas.

Key-Words: Arsenic, zinc, copper, bioaccumulation, groundwater, milkfish (Chanos chanos)

## **1** Introduction

Trace elements, such as arsenic (As), zinc (Zn) and copper (Cu), are introduced into the environment by a wide spectrum of natural and anthropogenic sources. These elements are non-biodegradable, and once they enter organisms the environment, from bioconcentration may occur in tissue by means of metabolic and biosorption processes [1, 6, 23]. Many studies have shown that trace elements can be accumulated in fish [22]. Uptake of sublethal concentrations of these elements may lead to altered physiological processes, which reduce the normal functioning of the organism. From an environmental point of view, a study on bioconcentration is important because elements usually occur in low concentrations and subtle physiological effects go unnoticed until gross chronic reactions (e.g. changes in population structure, altered reproduction, etc.) become

#### apparent.

Among trace elements, As is well known as a toxin [20]. Arsenic has been classified as a carcinogen, based on human epidemiological data; it is known to increase the risk of producing or inciting cancer [16]. Since As is mainly transported by water, it can easily be accumulated by aquatic organisms [12, 16, 18]. Several studies have been conducted and demonstrated that an overexposure of As could result in accumulation in fish and lead to adverse health effects [5, 8, 10]. Consumption of As-polluted fish might cause an overexposure of As and pose a cancer risk to human health via the food chain [9]. It has also been well documented that As is the major risk factor for blackfoot disease (BFD), which is a peripheral vascular disease that ends with dry gangrene and spontaneous amputation of affected extremities The increase in internal organ and skin [2].

cancers as well as BFD disease was significantly associated with the use of high-As groundwater [16].

Although Zn and Cu are essential nutrients for normal metabolic functioning [13], they may become toxic to aquatic organisms, particularly fish, when ambient concentrations exceed physiological thresholds [21, 24]. These elements can be incorporated into the food chain and concentrated by aquatic organisms to a level that affects their physiological state [7]. Ultimately they pose a health hazard to humans.

Many studies have demonstrated a modifying influence of water quality on Zn and Cu toxicity to fish [15]. For example, pH affects Zn and Cu speciation, which in turn affects bioavailability [3]; calcium (Ca<sup>2+</sup>) associated with water hardness tends to reduce toxicity by competitively inhibiting Cu binding to fish gills [17]; and increasing concentrations of dissolved organic matter sequester waterborne Cu from biological uptake [4, 14, 19].

Milkfish (Chanos chanos) is common seafood in Taiwan. Most of the milkfish culture ponds are located in the coastal region of southwest Taiwan, which is subjected to groundwater polluted with As [11]. A high amount  $(38,000-49,000 \text{ ton } \text{ha}^{-1})$  of freshwater is needed for milkfish culture. Several studies have been conducted to demonstrate that to use the groundwater for aquaculture may cause an overexposure of As [8, 9, 10]. In addition, milkfish from the aquacultural ponds in this area are also contaminated by Zn and Cu from the ambient water. The purpose of this study is to evaluate the bioaccumulation of As, Zn and Cu The bioconcentration factor of in milkfish. milkfish was estimated to analyze to potential risk to human health.

#### 2 Materials and Methods 2.1 Sampling

Samples of juvenile milkfish (range 4.0-6.0 cm in length and 0.41-1.41 g in weight) and ambient water were obtained from 8 culture ponds near the urban area. Three fish and three 500 ml water samples per pond were collected. The milkfish were placed on ice immediately, and kept at  $4^{\circ}$ C during transfer to the laboratory. The water samples were fixed by adding 5 ml 1N HNO<sub>3</sub>.

#### 2.2 Chemical Analysis

The frozen flesh of milkfish was dehydrated in a dryer (40) for 96 h, and then grounded into powder. Aliquots of dry flesh powder weighing 0.5 g were placed into a 250 ml beaker. Nitric acid (65%, 10 ml) was added for an overnight digestion. The beaker with flesh solution, after the digestion, was heated with a water bath at 70-80 for 2-4 h until the total volume is reduced to 1-2 ml. The solution was transferred to a volumetric flask (50 ml), and then filled with 0.01N of HNO<sub>3</sub> to make a 50 ml of final solution. After filtration, this 50 ml solution was transferred to test tubes for As, Zn and Cu analysis using ICP-MS (Agilent 7500a). Analytical quality control was achieved by digesting and analyzing identical amounts of rehydrated (90% H<sub>2</sub>O) standard reference materials (DORM-2, Dogfish Liver-2-organic matrix, NRC-CNRC, Canada). Recovery rates were ranged from 95% to 97%.

# **2.3** Calculation of Bioconcentration factor (BCF)

When steady-state chemical concentrations of tissue are attained, the equilibrium bioconcentration factor (BCF) of the milkfish can be calculated from the ratio of the chemical concentration in fish to that in water. The BCF can also be calculated from the ratio of the uptake rate constant to the depuration rate constant as,

$$BCF = \frac{C_b}{C_w} \tag{1}$$

where  $C_b$  (µg g<sup>-1</sup>) is the chemical concentration in biota;  $C_w$  (µg ml<sup>-1</sup>) is the chemical concentration in water.

#### **3** Results and Discussion

The resulting data of this study showed that the As, Cu, Zn concentrations in pond water were

 $63.92 \pm 57.71 \ \mu g \ L^{-1}, \ 69.36 \pm 27.81 \ \mu g \ L^{-1}$  and  $11.09 \pm 15.22 \ \mu g \ L^{-1}$ , respectively (Table 1), while in fish were  $0.94 \pm 1.34 \ \mu g \ g^{-1}, \ 2.01 \pm 0.96 \ \mu g \ g^{-1}$  and  $40.31 \pm 17.25 \ \mu g \ g^{-1}$ , respectively (Table 2).

The As, Zn and Cu levels in milkfish showed significant positive relations to the As, Zn and Cu concentrations in pond water:  $C_b = -0.27 +$ 0.02  $C_{\rm w}$ ,  $C_{\rm b} = 25.43 + 1.02 C_{\rm w}$  and  $C_{\rm b} = 1.22 + 1.02 C_{\rm w}$ 0.01  $C_{\rm w}$ , respectively (Fig. 1), where  $C_b$  is the chemical concentration in biota ( $\mu g g^{-1}$ );  $C_w$  is the chemical concentration in water ( $\mu g m l^{-1}$ ). It shows that milkfish can accumulate As, Zn and Cu from the ambient water. The high BCF values of As, Cu and Zn accumulated in fish  $(12.60 \pm 4.68, 32.31 \pm 12.53 \text{ and } 4029.04 \pm$ 1623.96, respectively) show that those cultured milkfish from are contaminated by the ambient water and have a high tolerance against the pollutants (Table 3). The accumulation of these chemicals may contribute to chronic toxicity in fish [1, 4]. Since those fish will be sold to the markets in cities nearby, ingestion of As, Zn and/or Cu contaminated milkfish could result in overexposure in inhabitants and lead to adverse health effects. Besides the risk on human health. using groundwater for aquaculture may also have a negative effect on the land above the aquifer because it may sink and cause damage on property and fields in the suburban and/or urban areas.

Lin et al. [9, 11] demonstrated that consumption of cultured milkfish from the As-contaminated ponds may pose a cancer risk Thus far no adverse effect on to human health. health of the people due to exposure to As, Zn and Cu contaminated milkfish is evidenced. A wider study involving As, Zn and Cu analyses of milkfish from unpolluted areas, as well as other aquacultural products should be investigated to assess the extent of chemical contamination in seafood. The interactions among the elements and the subsequent uptake and depuration rates associated with the individual elements are needed to be analyzed [4]. Further more, studies concerning the effects of the individual elements in the mixture are needed to be undertaken [23].

### 4 Conclusion

Milkfish can accumulate remarkably high levels of As, Zn and Cu from the ambient water. The As, Zn and Cu levels in milkfish showed significant positive relations to the As, Zn and Cu concentrations in pond water, respectively. The high BCF values of As, Zn and Cu accumulated in fish show that those cultured milkfish are contaminated by the ambient water and have a high tolerance against the pollutants. Since those fish will be sold to the markets in cities nearby, ingestion those contaminated milkfish could result in overexposure of As, Zn and/or Cu in inhabitants and lead to adverse health effects. Besides the risk on human health, using groundwater for aquaculture may also have a negative effect on the land above the aquifer because it may sink and cause damage on property and fields in the suburban and/or urban areas.

#### Reference:

- [1] Carpené, E, Cattani, O., Serrazaneti, G.P., Fedrizzi, G., Cortesi, P., Zinc and copper in fish from natural waters and rearing ponds in Northern Italy. *J. Fish Biol.* Vol.37, 1990, pp. 293-299.
- [2] Chiou, H.Y., Hsueh, Y.M., Liaw, K.F., Horng, S.F., Chiang, M.H., Pu, Y.S., Lin, J.S.N., Huang, C.H., Chen, C.J., Incidence of internal cancers and ingested inorganic As: a seven-year follow-up study in Taiwan. *Cancer Res.* Vol.55, 1995, pp. 1296-1300.
- [3] Cusimano, R.F., Brakke, D.F., Chapman, G.A., Effects of pH on the toxicities of cadmium, copper, and zinc to steelhead trout (*Salmo gairdneri*). *Can. J. Fish. Aquat. Sci.* Vol.43, 1985, pp. 1497-1503.
- [4] Dethloff, G.M., Schleck, D., Hamm, J.T., Bailey, H.C., Alterations in the physiological parameters of rainbow trout (Oncorhynchus mykiss) with exposure to copper and copper/zinc mixtures. *Ecotoxicol. Environ. Safety* Vol.42, 1999, pp. 253-254.
- [5] Donohue, J.M., and Abernathy, C.O., Exposure to inorganic arsenic from fish and shellfish. In: Chappell, W.R., Abernathy,

C.O., Calderon, R.L. (Eds.), Arsenic exposure and health effects, Elsevier, Oxford, 1999, p. 89-98.

- [6] Hodson, P.V., The effect of metal metabolism on uptake, disposition and toxicity in fish. *Aquat. Toxicol.* Vol.11, 1988, pp. 3-18.
- [7] Liao, C.M., Chen, B.C., Lin, M.C., Chiu, H.M., Chou, Y.H. Coupling toxicokinetics and pharmacodynamics for predicting survival of abalone (*Haliotis diversicolor supertexta*) exposed to waterborne zinc. *Environ. Toxicol.* Vol.17, 2002, pp. 478-486.
- [8] Liao, C.M., Chen, B.C., Singh S., Lin, M.C., Liu, C.W., Han, B.C.. Acute toxicity and bioaccumulation of arsenic in tilapia *Oceochromis mossambicus* from blackfoot disease area in Taiwan. *Environ. Toxicol.* Vol.18, 2003, pp. 252-259.
- [9] Lin, M.C., Cheng, H.H., Lin, H.Y., Chen, Y.C., Chen Y.P., Liao, C.M., Chang-Chien, G.P., Dai, C.F., Han, B.C., Liu, C.W., Arsenic accumulation and acute toxicity in milkfish (*Chanos chanos*) from blackfoot disease area in Taiwan. *Bull. Environ. Contam. Toxicol.* Vol.72, 2004, pp. 248-254.
- [10] Lin, M.C., Liao, C.M., Liu, C.W., Singh, S., Bioaccumulation of arsenic in aquacultural large-scale mullet Liza macrolepis from the blackfoot disease area in Taiwan. *Bull. Environ. Contam. Toxicol.* Vol.67, 2001, pp. 91-97.
- [11] Lin, M.C., Lin, S.Y., Risk assessment of arsenic exposure caused by consumption of cultured milkfish (*Chanos chanos*) from blackfoot disease area in Taiwan. Aquaculture. *Bull. Environ. Contam. Toxicol.* Vol.75, 2005, (in press).
- [12] Liu, C.W., Lin, K.H., Kuo, Y.M., Application of factor analysis in the assessment of groundwater quality in a blackfoot disease area in Taiwan. *Sci. Total Environ.* Vol.313, 2003, pp. 77-89.
- [13] Mertz, W., The essential trace elements. *Science* Vol.213, 1981, pp. 1332-1338.
- [14] Mazon, A.F., Fernandes, M.N., Toxicity and differential tissue accumulation of

copper in the tropical freshwater fish, *Prochilodus scrofa* (Prochilodontidae). *Bull. Environ. Contam. Toxicol.* Vol.63, 1999, pp. 797-804.

- [15] Mukhopadhyay, M.K., Konar, S.K., Effects of copper, zinc and iron mixture on fish and aquatic ecosystem. *Environ. Ecol.* Vol.3, 1985, pp. 58-64.
- [16] Ng, J.C., Wang, J.P., Shraim A., A global health problem caused by arsenic from natural sources. *Chemosphere* Vol.52, 2003, pp. 1353-1359.
- [17] Pagenkopf, G. K., Gill surface interaction model for trace-metal toxicity to fishes: role of complexation, pH, and water hardness. *Environ. Sci. Technol.* Vol.17, 1983, pp. 342-347.
- [18] Phillips, D. J. H., Arsenic in aquatic organisms: a review, emphasizing chemical speciation. *Aquat. Toxicol.* Vol.16, 1990, pp. 151-186.
- [19] Playle, R.C., Gensemer, R.W., Dixon, D.G., Copper accumulation on gills of fathead minnows: influence of water hardness, complexation and pH of the gill micro-environment. *Environ. Toxicol. Chem.* Vol.11, 1992, pp. 381-391.
- [20] Thomas, D.J., Arsenic toxicity in humans: research problems and prospects. Environ. *Geochem. Health* Vol.16, 1994, pp. 107-111.
- [21] Vallee, B.L., Falchuk, K.H., The biochemical basis of zinc physiology. *Physiol. Rev.* Vol.73, 1993, pp. 79-118.
- [22] Villegas-Navarro, A., Villarreal-Trevino, C.M., Differential uptake of zinc, copper, and lead in Texas cichlid (*Cichlasoma cyanoguttatum*). *Bull. Environ. Contam. Toxicol.* Vol.42, 1989, pp. 761-768.
- [23] Wepener, V., Vuren, J.H.J., Preez, H.H., Uptake and distribution of a copper, iron and zinc mixture in gill, liver and plasma of a freshwater teleost, *Tilapia sparrmanii*. *Water SA* Vol.27, 2001, pp. 99-108.

Concentration in Water Mean ± SE			
As	Zn	Cu	
88.10 ± 65.63*	$21.65 \pm 11.06$	$85.70 \pm 8.34*$	
$102.57 \pm 63.62*$	$5.97 \pm 1.91$	$60.01 \pm 2.44*$	
$30.44\pm 6.82$	$7.45\pm0.23$	$40.78 \pm 15.92*$	
$34.58\pm3.76$	$8.49\pm0.23$	$121.88 \pm 7.18*$	
$63.92 \pm 57.71$	$11.90 \pm 15.22$	$69.36\pm27.81$	
	Conce As $88.10 \pm 65.63^*$ $102.57 \pm 63.62^*$ $30.44 \pm 6.82$ $34.58 \pm 3.76$ $63.92 \pm 57.71$	Concentration in Water MeanAsZn $88.10 \pm 65.63^*$ $21.65 \pm 11.06$ $102.57 \pm 63.62^*$ $5.97 \pm 1.91$ $30.44 \pm 6.82$ $7.45 \pm 0.23$ $34.58 \pm 3.76$ $8.49 \pm 0.23$ $63.92 \pm 57.71$ $11.90 \pm 15.22$	

Table 1. Arsenic (As), Zinc (Zn) and Copper (Cu) concentrations in water of milkfish culture ponds ( $\mu g L^{-1}$ ).

\* As > 50  $\mu$ g L<sup>-1</sup>; Zn > 500  $\mu$ g L<sup>-1</sup>; Cu > 30  $\mu$ g L<sup>-1</sup>

Table 2. Arsenic (As), Zinc (Zn) and Copper (Cu) levels in milkfish ( $\mu$ g L<sup>-1</sup>) from culture ponds.

Location –	Level in Fish Mean ± SE			
	As	Zn	Cu	
Putai	$1.89 \pm 1.66$	$46.33\pm13.75$	$2.11 \pm 0.03$	
Yichu	$1.29\pm0.93$	$33.10 \pm 4.61$	$1.84 \pm 0.11$	
Hsuehchia	$0.32\pm0.00$	$39.92\pm2.63$	$1.76 \pm 0.31$	
Peimen	$0.40\pm0.11$	$32.57\pm2.65$	$2.63 \pm 0.45$	
Average	$0.94 \pm 1.34$	$40.31 \pm 17.25$	$2.01 \pm 0.96$	

Table 3. Bioconcentration factors of Arsenic (As), Zinc (Zn) and Copper (Cu) of milkfish from culture ponds.

Location –		BCF	
	As	Zn	Cu
Putai	$16.85\pm9.72$	$3191.7 \pm 2404.89$	$24.84\pm2.92$
Yichu	$9.01\pm6.64$	$5541.02 \pm 1469.51$	$30.78\pm3.81$
Hsuehchia	$10.94\pm3.55$	$5357.59 \pm 2125.76$	$47.43 \pm 15.61$
Peimen	$10.35\pm4.17$	$3919.48 \pm 593.07$	$21.62 \pm 7.44$
Average	$12.51\pm4.70$	$4029.04 \pm 1623.96$	$32.31 \pm 12.53$



Figure 1. Plot of the As, Zn and Cu levels in cultured milkfish and the chemical concentrations in pond water.